

Sweet 16 Chemistry Compound Tournament Answers



Sweet 16 Chemistry Compound Tournament

With spring just around the corner, your students' thoughts will soon be turning to sunshine, prom, and the NCAA basketball tournament. This clever activity combines the ever-popular March Madness basketball pool with a review of chemical formulas, the structure and charges of ions, and molar mass calculations. The result is the Sweet 16 Chemistry Compound Tournament.

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Sweet 16 Chemistry Ion Tournament - Chemistry Ms. McChesney

Sweet 16 Chemistry Ion Tournament First Round Second Round Semifinals Finals Winner NiCl₂ Ni(II)²⁺ Fe(II) S Zn CO₃ Na CrO₄ Mg PO₄ Ag(I) SO₄ Ba NO₃ Al OH Rules 1. First Round →Add Charges 2. Second Round →Write Formula 3. Semis →Soluble Compound (compound that will dissolve in water) 4. Finals →Soluble Product of Chemical Reaction 5.

Name Class/Period Sweet 16 Chemistry Ion Tournament

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Sweet 16 Chemistry Compound Tournament. First Round Second Round SemifinalsFinalsWinner.

Sodium Acetate. Barium Phosphate. Potassium Carbonate. Zinc Chloride. Ammonium Sulfate. Lithium Carbonate. Potassium Cyanide. Aluminum Nitrate. Mercury (I) Sulfate. Titanium (IV) Oxide. Iron (III) Sulfate. Sodium Bicarbonate. RULES Round 1 - Write formula.

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Potassium phosphate First Round Second Round Semifinals Finals Winner Rules 1. First Round — Write formula, most number of atoms in formula unit wins 2. Second Round — Greater number of ions in the formula unit wins 3. Semis — Higher charge on anion wins 4. Finals — The larger molar mass wins Sweet 16 Chemistry Compound Tournament

Potassium phosphate First Round Second Round Semifinals ...

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combines the ever-popular basketball office pool with a review of the formulas of ionic compounds and their solubility.

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